

# RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College affiliated to University of Calcutta)

B.A./B.Sc. SIXTH SEMESTER EXAMINATION, JUNE 2022

THIRD YEAR (BATCH 2019-22)

CHEMISTRY (HONOURS)

Date : 17/06/2022

Time : 11.00 am – 1.00 pm

Paper : DSE-3

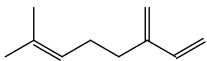
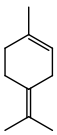
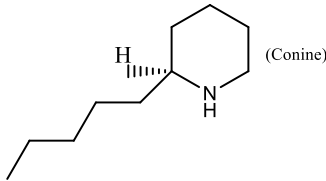
Full Marks : 50

Attempt **any five** questions :

1. a) E, F and G are the three aldohexoses. E and F yield D-sorbitol when they are Catalytically hydrogenated. E and F yield different osazones when treated with excess phenylhydrazine. F and G give the same osazone but different alditols. Give the structures of E, F and G assuming that F and G are D-aldohexoses. [3]  
b) Predict the products (A) to (C) in the following reactions. [3]  
i) Methyl  $\alpha$ -D-glucofuranoside  $\xrightarrow[2\text{-equivalent}]{\text{HIO}_4}$  (A)  $\xrightarrow[\text{H}^+]{\text{H}_2\text{O}}$  (B)  
ii) Methyl  $\alpha$ -D-glucopyranoside  $\xrightarrow[\text{H}^+]{\text{PhCHO}}$  (C)  
c) Convert :  $\alpha$ -D-glucopyranose  $\rightarrow$   $\alpha$ -3-Deoxy-D-glucopyranose [2]  
d) Utilising the structural illustration, explain the non-reducing property of sucrose. [2]
2. a) Give the mechanism of osazone formation from D-glucose. Cite one evidence in favour of the mechanism. why osazone formation is stopped at the second carbon? Justify your answer. [3]  
b) Predict the products of the following reactions and explain their formation. [2 $\times$ 2]  
i) D-Arabinose  $\xrightarrow[\text{high pH}]{\text{HCN}}$   
ii)  $\alpha$ -D-glucopyranose  $\xrightarrow[2) \text{Ph} \text{---} \text{Cl} | \text{base}]{1) \text{Me}_2\text{CO}/\text{H}^+}$   
c) The concept of 'end group interchange' was used by Fischer to distinguish the configurational structure of D-glucose and D-mannose. Explain. [2]  
d) Write the different interactions responsible for stabilising the secondary structure of a protein. [1]
3. a) Justify or Criticise : [2 $\times$ 3]  
i)  $4c_1$  conformation of D-glucose is more stable than its  $1c_4$  conformation whereas  $1c_4$  conformation of L-galactose is more stable than its  $4c_1$  conformation.  
ii) Formation of D-mannonitrile from D-arabinose is a thermodynamically controlled reaction.  
iii) Mutarotation of D-glucose is a case of first order asymmetric transformation which is not true for mutarotation of D-gluconolactone.  
b) Population distribution of  $\alpha$ - and  $\beta$ -anomer of D-glucose and D-mannose is as follows : [4]

	$\alpha$ -anomer	$\beta$ -anomer	Medium
D-glucose	36%	64%	H <sub>2</sub> O
D-mannose	69%	31%	H <sub>2</sub> O

Explain this observation.

4. a) How is N-terminal amino acid determined by Edman's method? [2]  
 b) Describe the denaturation of protein by  
 i) changing pH, ii) heavy metal( $Pb^{+2}$ ) addition. [2]  
 c) Discuss the role of CNBr in peptide sequence analysis. [2]  
 d) Denaturation of protein cannot disrupt the primary structure of a protein. [True/False] [1]  
 e) Distinguish Lys.gly and gly.Lys. [2]  
 f) How would you explain the reducing property of D-fructose? [1]
5. a) What is post-translation modification? [2]  
 b) Write down the biological role for [3×2]  
 (i) RNA polymerase.  
 (ii) mRNA  
 (iii) tRNA  
 c) Write down the schematic diagram for DNA → protein synthesis. [2]
6. a) What are Nucleosides and Nucleotides? Give example of each. [2]  
 b) Draw the hydrogen bonding present in DNA shown in Watson-crick model. [3]  
 c) Draw the structure of purine and pyrimidine bases that are present in RNA. [2]  
 d) i) Write down the process of DNA replication in schematic diagram. [2]  
 ii) Why DNA is called Genetic code of the life. [1]
7. a) Write down the class, structure and the medical importance or biological effect of the following alkaloids : [3]  
 (i) Nicotine  
 (ii) Cocaine  
 (iii) ephedrine  
 b) i) What is the monomeric unit for Terpenes? [1]  
 ii) Explain the isoprene rule with example. [2]  
 c) Give industrial preparation of Citral. [2]  
 d) What are terpinoids? Give one example. [2]
8. a) What are sesquiterpenes and diterpenes. Give one example of each. [4]  
 b) Find out the head-tail connection in the following terpenes :- [2]  
 (i)   
 (ii)   
 c) Give Schematic presentation for isolation of alkaloids from nature. [2]  
 d) The following molecule conine is found in hemlock poison. What class/type of alkaloids in this : [1]  
 (Conine)  
 e) Give one example and biological effect of indole type alkaloids. [1]

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